

Michaelis-Menten reaktionskinetik

> restart

https://en.wikipedia.org/wiki/Michaelis%E2%80%93Menten_kinetics#Derivation

Differentialligningssystemet fra Wikipedia:

$$\begin{aligned} d[S]/dt &= -k_f[E][S] + k_r[ES] \\ d[E]/dt &= -k_f[E][S] + k_r[ES] + k_{cat}[ES] \\ d[ES]/dt &= +k_f[E][S] - k_r[ES] - k_{cat}[ES] \\ d[P]/dt &= +k_{cat}[ES] \end{aligned}$$

Differentialligningssystemet opskrevet i matematisk notation:

$$\begin{aligned} S'(t) &= -k_f \cdot E(t) \cdot S(t) + k_r \cdot ES(t) \\ E'(t) &= -k_f \cdot E(t) \cdot S(t) + k_r \cdot ES(t) + k_{cat} \cdot ES(t) \\ ES'(t) &= +k_f \cdot E(t) \cdot S(t) - k_r \cdot ES(t) - k_{cat} \cdot ES(t) \\ P'(t) &= +k_{cat} \cdot ES(t) \end{aligned}$$

Differentialligningssystemet defineres i Maple.
4 differentialligninger med 4 funktioner:

$$\begin{aligned} > DL1 := S'(t) = -k_f \cdot E(t) \cdot S(t) + k_r \cdot ES(t) \\ & \quad DL1 := D(S)(t) = -k_f E(t) S(t) + k_r ES(t) \end{aligned} \quad (1)$$

$$\begin{aligned} > DL2 := E'(t) = -k_f \cdot E(t) \cdot S(t) + k_r \cdot ES(t) + k_{cat} \cdot ES(t) \\ & \quad DL2 := D(E)(t) = -k_f E(t) S(t) + k_r ES(t) + k_{cat} ES(t) \end{aligned} \quad (2)$$

$$\begin{aligned} > DL3 := ES'(t) = +k_f \cdot E(t) \cdot S(t) - k_r \cdot ES(t) - k_{cat} \cdot ES(t) \\ & \quad DL3 := D(ES)(t) = k_f E(t) S(t) - k_r ES(t) - k_{cat} ES(t) \end{aligned} \quad (3)$$

$$\begin{aligned} > DL4 := P'(t) = +k_{cat} \cdot ES(t) \\ & \quad DL4 := D(P)(t) = k_{cat} ES(t) \end{aligned} \quad (4)$$

Begyndelsesbetingelser:

http://se.wtb.tue.nl/sewiki/biological_systems/de#michaelis-menten_kinetics

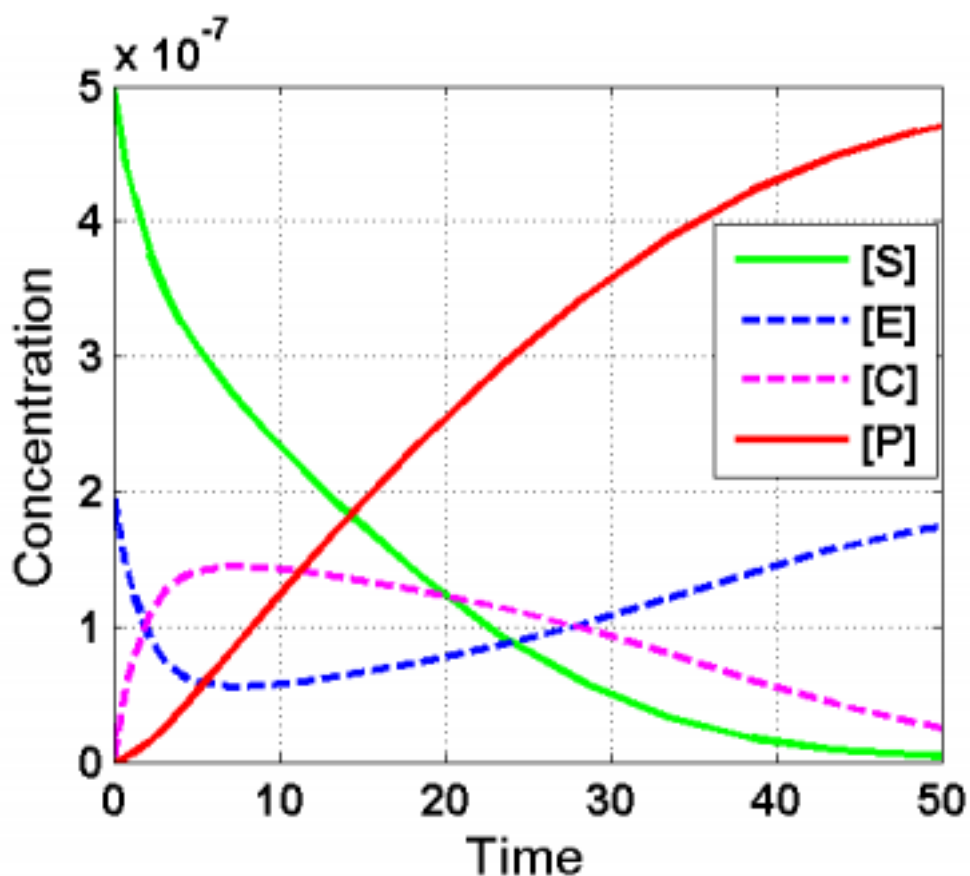
NB: link virker IKKE!

- $S(0) = 5e^{-7}$,
- $E(0) = 2e^{-7}$,
- $C(0) = 0$,
- $P(0) = 0$,
- $k_1 = 1e^6$,
- $k_2 = 1e^{-4}$,
- $k_3 = 0.1$.

Forventet graf:

http://se.wtb.tue.nl/sewiki/biological_systems/de#michaelis-menten_kinetics

NB: link virker IKKE!



> $k_f := 1 \cdot 10^6$; $k_r := 1 \cdot 10^{-4}$; $k_{cat} := 0.1$

$$k_f := 1000000$$

$$k_r := \frac{1}{10000}$$

$$k_{cat} := 0.1$$

(5)

Numerisk løsning af differentiaalligningssystemet:

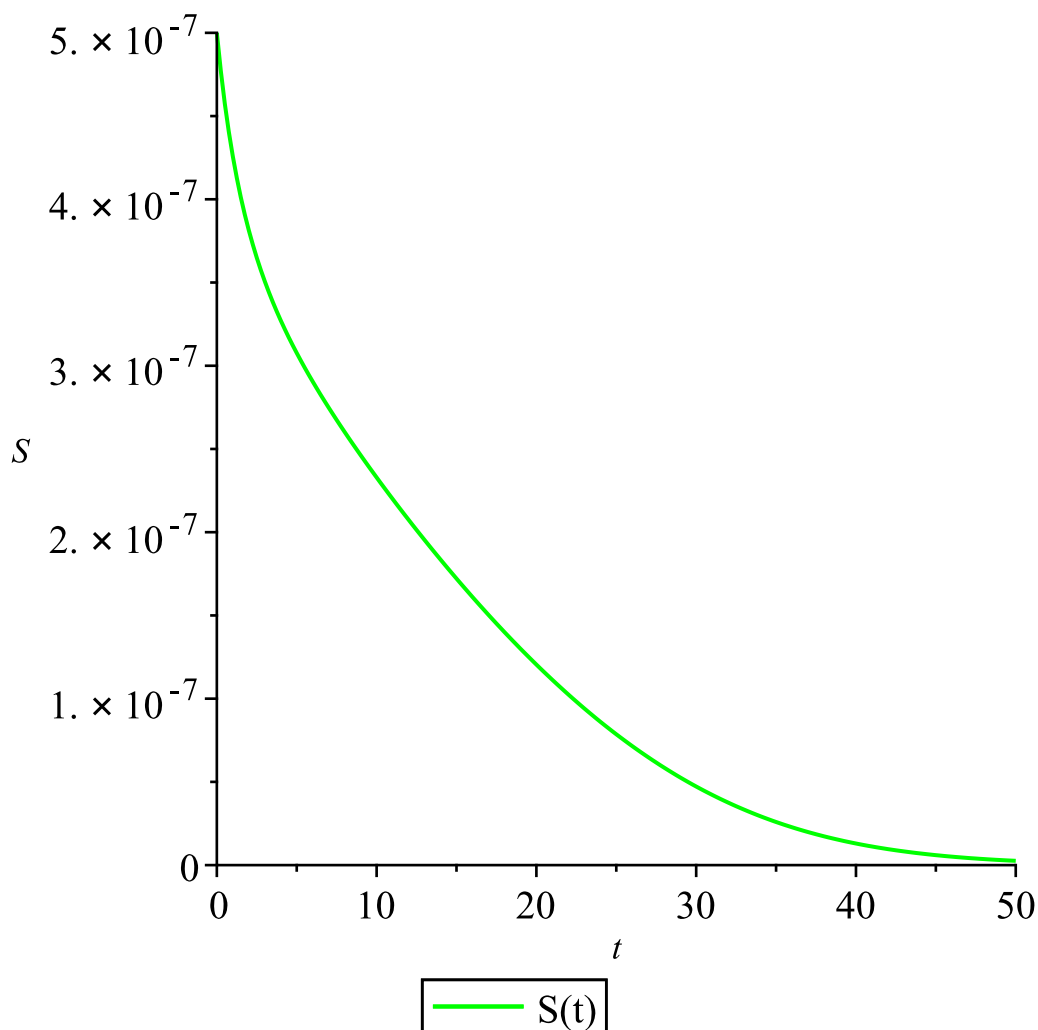
https://en.wikipedia.org/wiki/Numerical_methods_for_ordinary_differential_equations

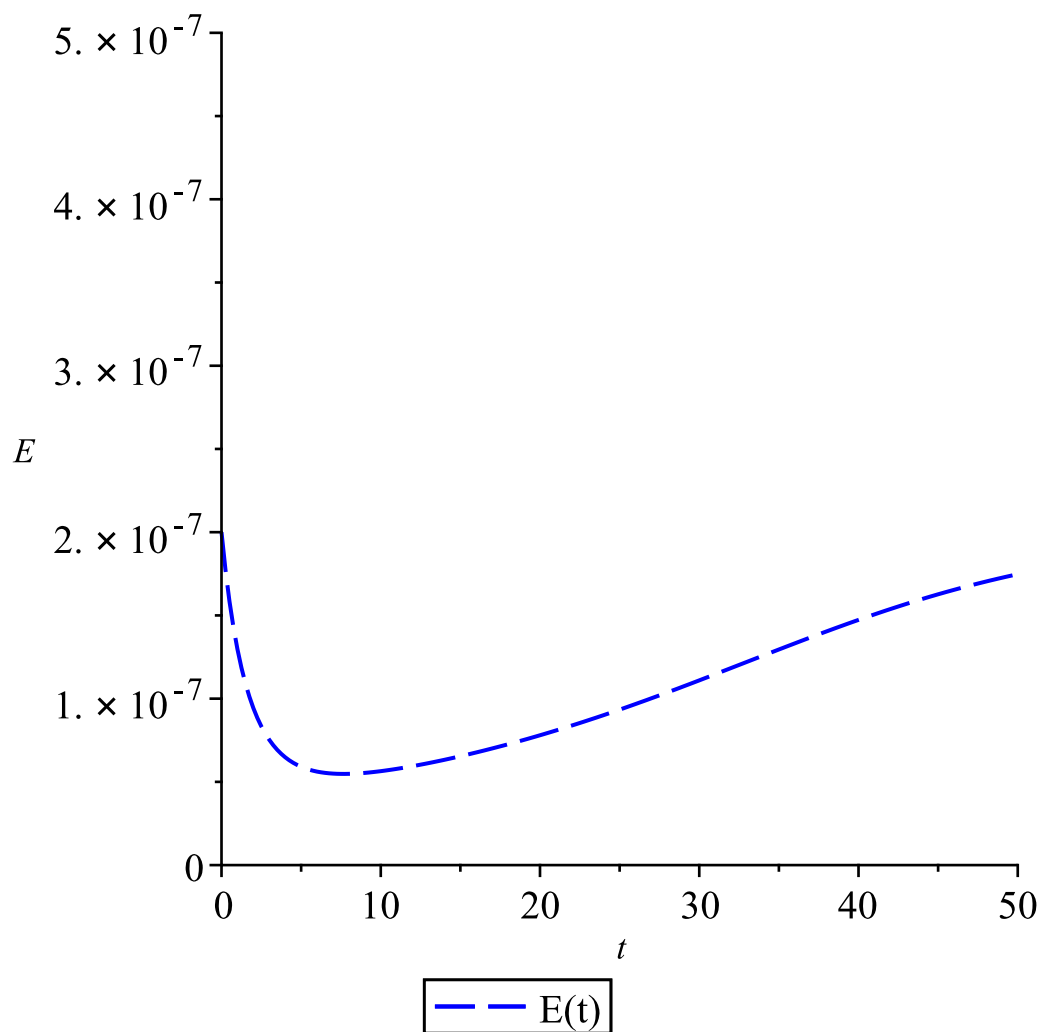
https://en.wikipedia.org/wiki/Runge%E2%80%93Kutta_methods

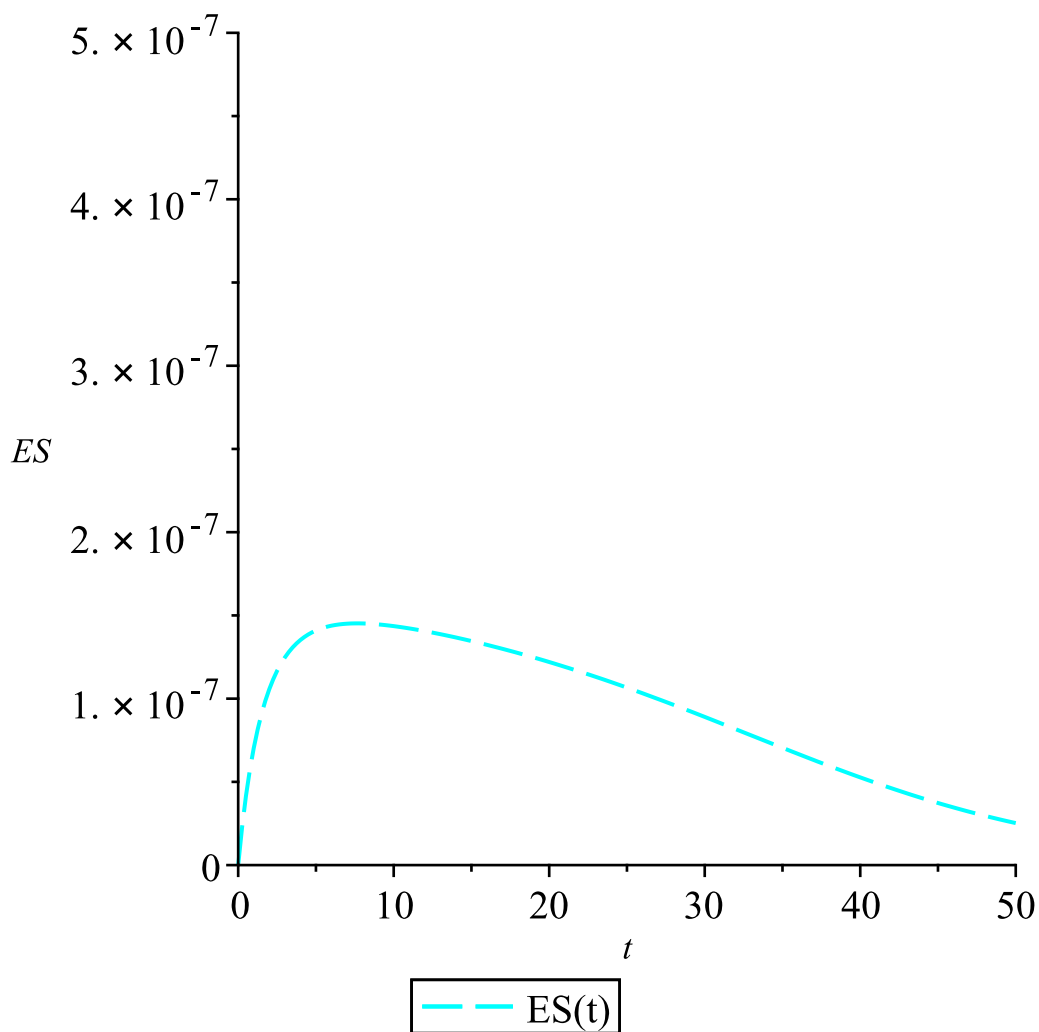
https://en.wikipedia.org/wiki/Runge%E2%80%93Kutta%E2%80%93Fehlberg_method

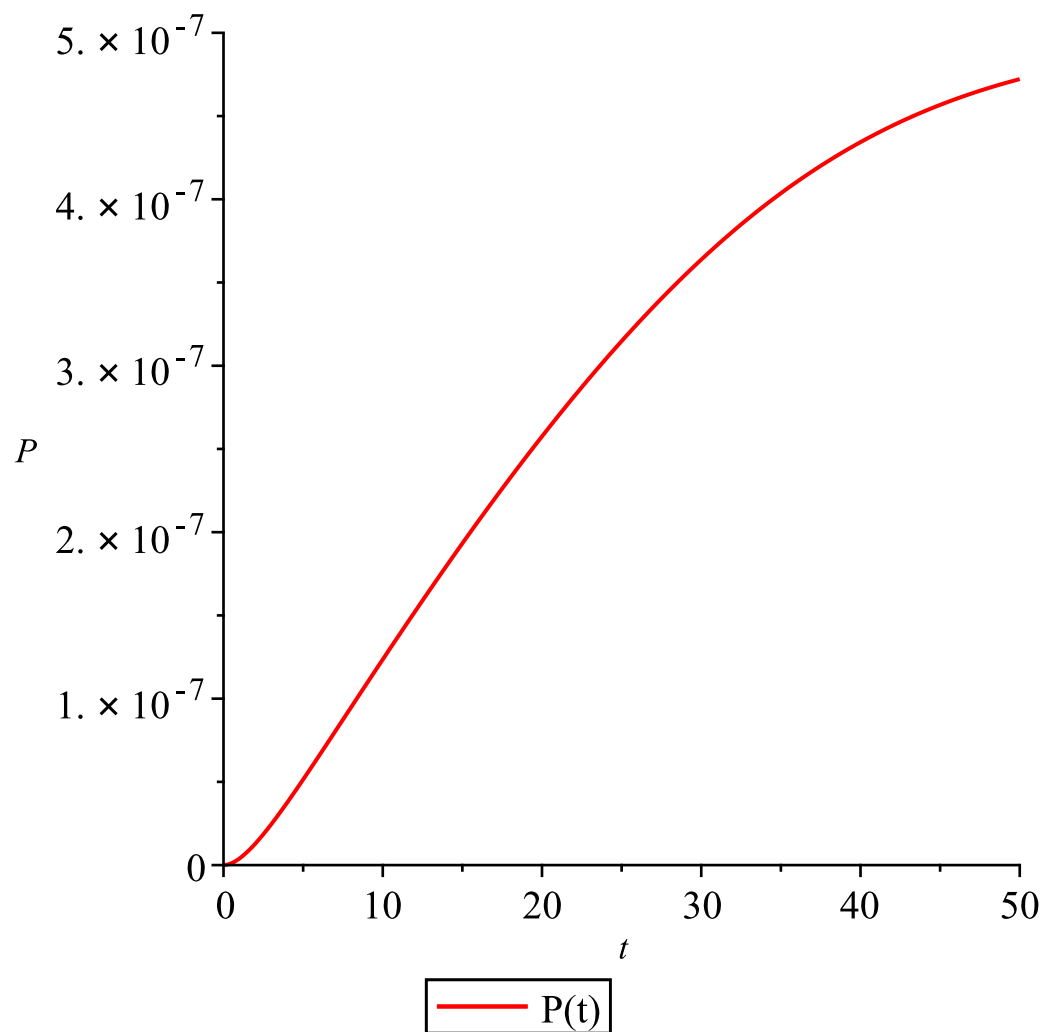
Differentialligningssystemet løses numerisk (tilnærmet), da det ikke kan løses eksakt:

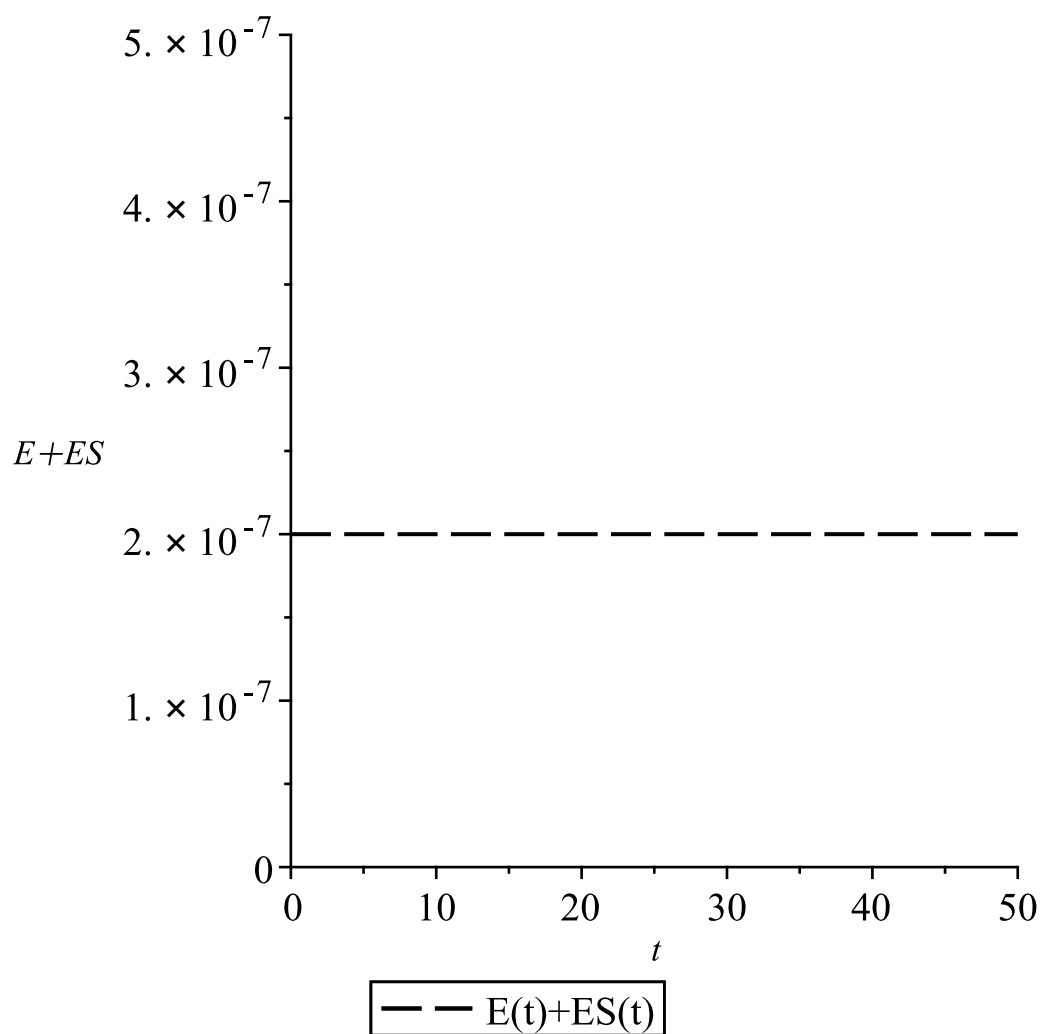
```
> løsning := dsolve( {DL1, DL2, DL3, DL4, S(0) = 5·10-7, E(0) = 2·10-7, ES(0) = 0, P(0) = 0}, numeric, method = rkf45, abserr = 10-10 );
> with( plots ) :
> odeplot( løsning, [t, S(t)], t = 0 .. 50, color = green, legend = "S(t)" ) : plotS := % :
  display( plotS, view = [0 .. 50, 0 .. 5·10-7] );
odeplot( løsning, [t, E(t)], t = 0 .. 50, color = blue, linestyle = dash, legend = "E(t)" ) : plotE :=
  % : display( plotE, view = [0 .. 50, 0 .. 5·10-7] );
odeplot( løsning, [t, ES(t)], t = 0 .. 50, color = cyan, linestyle = dash, legend = "ES(t)" ) :
  plotES := % : display( plotES, view = [0 .. 50, 0 .. 5·10-7] );
odeplot( løsning, [t, P(t)], t = 0 .. 50, color = red, legend = "P(t)" ) : plotP := % :
  display( plotP, view = [0 .. 50, 0 .. 5·10-7] );
odeplot( løsning, [t, E(t) + ES(t)], t = 0 .. 50, color = black, linestyle = dash, legend
  = "E(t)+ES(t)" ) : plotEES := % : display( plotEES, view = [0 .. 50, 0 .. 5·10-7] );
```



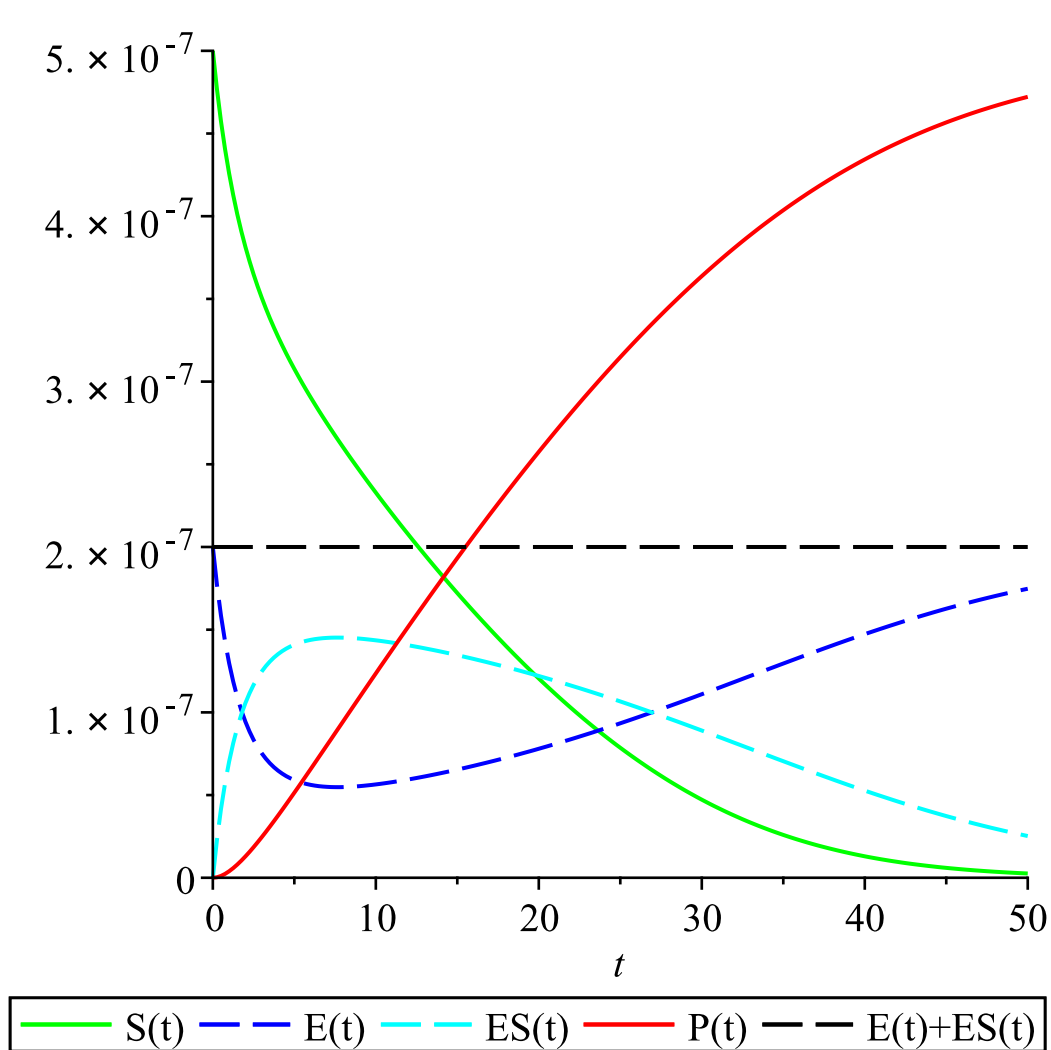








```
> display(plotS, plotE, plotES, plotP, plotEES, view=[0..50, 0..5·10-7], labels=[t, ""])
```



Farver på den samlede graf:

GRØN = [S]

RØD = [P]

CYAN = [ES]

BLÅ = [E]

SORT = [E] + [ES]

*NB: [E] + [ES] er konstant, dvs. enzymet forbruges ikke.
Substratet [S] forbruges, men produktet [P] fremstilles.*